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phases we hold onto in phrases like "gin and tonic," where the tonic is dissolved in the gin). Using that, we wind up with two more descriptions of the situation above: Side 1 is hypertonic to side 2. That means that the solution on side 1 has more solute

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dissolved in it than
the solution on side
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Answer Sheet

OSMOSIS is the diffusion of water through a cell membrane. When water goes through the cell membrane, no matter what direction, in or out, it is called

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"osmosis". In class we observed osmosis in plant cell by: FIRST putting a drop of water on a slide.
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is a special type of diffusion. It is the diffusion of solvent or water across a semi-permeable membrane (a membrane that allows for the diffusion of certain solutes and water) from an area of higher

Diffusion And

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Osmosis is the diffusion of water across a membrane in response to osmotic pressure caused by an imbalance of molecules on either side of the membrane.

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Osmoregulation is the process of maintenance of salt and water balance (osmotic balance) across membranes within the body's fluids, which are composed of water plus electrolytes and non ...

Osmoregulation

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Concentration of a
solute on either
side of a

membrane in a cell
and observe the
system as it
adjusts to the
conditions through
osmosis. The initial
concentration of
the solute can be
manipulated, along
with the volume of
the cell.

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Osmosis Gizmo :
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Osmosis. Imagine
you have a cup
that has (100 mL) water,
and you add (15 g) of table
sugar to the water.
The sugar dissolves
and the mixture
that is now in the
cup is made up of a

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solute (the sugar) that is dissolved in the solvent (the water). The mixture of a solute in a solvent is called a solution.. Imagine now that you have a second cup with (100 mL) of ...

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