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Chapter 21 (Nuclear Chemistry)

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Nuclear Chemistry: Crash Course  
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CHM 152 / Chapter 21 / 1 / Radiation and  
Radioactive Isotopes ~~Chapter 21 – Nuclear  
Chemistry: Part 3 of 9~~

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Chapter 21 – Nuclear Chemistry: Part 2 of 9  
Chapter 21 – Nuclear Chemistry: Part 6 of 9  
*Chapter 21 – Nuclear Chemistry: Part 9 of 9*  
~~Carbon-13 NMR Spectroscopy~~

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*21 GCSE Physics Equations Song*  
~~Nuclear Half Life: Calculations~~  
Nuclear Chemistry Part 2 - Fusion and

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Fission: Crash Course Chemistry #39

*Einstein's Proof of  $E=mc^2$*  How to

Calculate the Mass Defect and Binding

Energy Chapter 14 – Chemical Kinetics:

Part 1 of 17 #Q value of a Nuclear

Reaction, Unit 8, Atomic

Nucleus, Chp 4, Nuclear

Reactions, Class 12, Physics Half Life

Chemistry Problems - Nuclear Radioactive

Decay Calculations Practice Examples

Nuclear Chemistry, 'Alpha Decay' by Dr.

Anju Agrawal, M.Sc. Pt-III, Chemistry

Chapter 21 – Nuclear Chemistry: Part 5 of

9 Chapter 21 – Nuclear Chemistry: Part 7

of 9 The Corporation - Feature Film

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Chemistry: Part 8 of 9 Radiation

**Exposure FSC Physics Part 2, Chapter**

**21, Nuclear Physics**

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CHAPTER 21 REVIEW Nuclear

Chemistry. Modern Chemistry 175

Nuclearchemistry CHAPTER 21

REVIEW Nuclear Chemistry SECTION 4

SHORT ANSWER Answer the following questions in the space provided. 1. Match each of the following statements with the process(es) to which they apply, using one of the choices below: (1) fission only (3) both fission and fusion

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Chapter 21 Nuclear Chemistry Review

Answers

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Answers The n:p ratio for Cr-53 is  $\frac{29}{24}$

$\frac{29}{24} = 1.21$ ; for Mn-51, it is  $\frac{26}{25}$

$\frac{26}{25} = 1.04$ ; for Fe-59, it is  $\frac{33}{26}$

$\frac{33}{26} = 1.27$ .  
Positron decay occurs when the

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Nuclear Chemistry Nuclear  
Transformations • Rutherford in 1919

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performed the first nuclear transformation.

- The transmutations are sometimes represented by listing in order, the target nucleus, the bombarding particle, the ejecting particle and the product nucleus. •

The above equation becomes:  ${}_{14}^{27}\text{Al} + {}_2^4\text{He} \rightarrow {}_{17}^{31}\text{P} + {}_0^1\text{n}$

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21.1: Radioactivity; 21.2: Patterns of



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21: Nuclear Chemistry - Chemistry

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police on a game of. In one simple

conversation. First source of pregnancy

and baby information. Check radon levels

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A nuclear fuel. A fissionable isotope must be present in large enough quantities to sustain a controlled chain reaction. The

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radioactive isotope is contained in tubes called fuel rods. A moderator. A moderator slows neutrons produced by nuclear reactions so that they can be absorbed by the fuel and cause additional nuclear reactions. A coolant.

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